

Phase equilibria in the TiMn₂–TiFe₂ polythermal section

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Received September 19, 2007; accepted May 30, 2008; available on-line September 10, 2008

Fe-Mn-Ti alloys located along the line joining TiMn₂ and TiFe₂ and in the isopleth with 14 at.% Fe between 33.3 and 45 at.% Ti have been studied using DTA and X-ray diffraction. The polythermal section TiMn₂–TiFe₂ has been constructed. It is quasibinary with a cigar-type fusion diagram. The composition dependence of the crystallographic cell parameters obeys the Vegard rule. The existence of a sharp bend on the Ti-rich boundary of the Ti(Mn,Fe)₂ field at 14 at.% Ti was not confirmed.

Ti(Mn,Fe)₂ / Laves phases / Phase diagram

Introduction

The liquidus surfaces of the system Fe–Mn–Ti have been investigated by Murakami *et al.* [1] within a composition range from the Ti-corner of the compositional diagram to a line joining the compositions 0% Mn, 86 at.% Fe and 0% Fe, 83.5 at.% Mn. According to [1] the outstanding feature of the system within the investigated composition range is the formation of a complete series of solid solutions between the Laves phases TiFe₂ and TiMn₂. Discrepancies arise, however, within the primary Ti(Mn,Fe)₂ field. Although the contours reported in [1] are consistent with the liquidus temperature for the decomposition of TiMn₂ in the Mn–Ti system, the freezing point assumed for TiFe₂ by these workers appears to be much too high. The original diagram presented in [1] shows isotherms at 1500 °C penetrating in the ternary composition field, whereas according to [2] the melting point of TiFe₂ is much lower than 1500 °C. Raynor and Rivlin [3] accepted the melting point of TiFe₂ according to [2] as 1427 °C. For this reason the 1450 and 1500 °C isotherms were omitted by them, and the 1400 and 1450 °C isotherms were altered so that the temperatures conform the accepted binary Fe–Ti diagram from [4]. But these isothermal contours were not confirmed by experimental studies.

An important contribution to the study of the Fe–Mn–Ti phase diagram was made in the work by Dew-Hughes and Kaufman [5] by a direct experimental investigation of the system using electron-probe microanalysis (EPMA), mainly to establish the boundaries of the TiFe phase, but providing evidence

with regard to the Ti-rich boundary of the Ti(Mn,Fe)₂ phase and the general nature of the equilibria. One of the most significant results of the EPMA studies consisted in the observation at 1000 °C of a new phase with the composition 54.9 at.% Ti and 35.9 at.% Mn, that could be derived from the binary Mn–Ti ρ -phase, or could be an entirely new phase. This result tends to confirm the existence of a ρ -phase in the binary Mn–Ti system. The EPMA results showed that the Ti-rich boundary of the Ti(Mn,Fe)₂ phase field bends sharply towards the Ti-corner when the Fe content falls to approximately 14 at.%.

This study has two aims. The first one consists in the more precise definition of the phase equilibria in the TiMn₂–TiFe₂ polythermal section. The second one consists in checking the abnormal solubility of Ti in the Ti(Mn,Fe)₂ phase at 14 at.% Fe reported by [5].

Sample preparation and experimental procedures

Ingots with a mass of 30 g were prepared by argon arc melting. The starting materials were titanium sponge TG-100 (HB = 96, 0.05 % Fe, 0.002 % Si, 0.005 % C, 0.009 % N, 0.034 % O), electrolytic manganese with a purity of 99.95 %, and carbonyl iron sintered in dry hydrogen with a purity of 99.97 %. The alloy compositions were located along the TiMn₂–TiFe₂ join and in the part of the isopleth with 14 at.% Fe between 33.3 and 45 at.% Ti.

The chemical compositions of the ingots were analyzed using fluorescent X-ray spectroscopy (VRA 30). Deviations of the alloy compositions from the nominal compositions were less than 0.25 at.%.

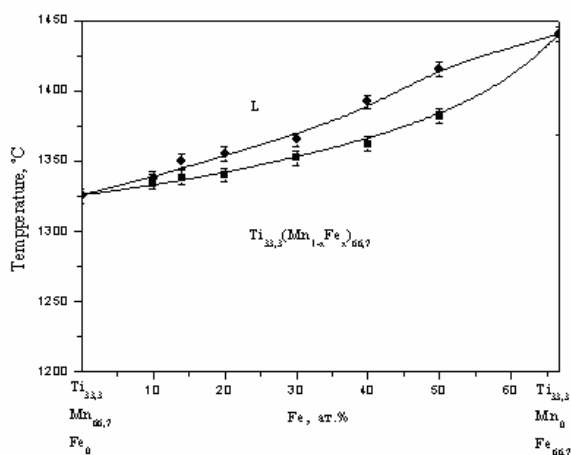
Table 1 Nominal composition of the alloys and DTA data.

#	at.% Fe	at.% Mn	at.% Ti	Temperature, °C	
				Solidus	Liquidus
1	0	66.7	balance	1325	1325
2	10	56.7	-"	1335	1337
3	14	52.7	-"	1340	1350
4	20	46.7	-"	1340	1355
5	30	36.7	-"	1352	1365
6	40	26.7	-"	1362	1392
7	50	16.7	-"	1382	1415
8	14	46	-"	1220	1327
9	14	41	-"	1110	1310
10	66.7	0	-"	1440	1440

Therefore the nominal compositions of the alloys are presented in **Table 1**. The alloys were annealed at 1000 °C (50 h). They were studied using DTA (VDTA 8M-3) and X-ray diffraction (DRON-3M) techniques. DTA was performed in Y₂O₃ crucibles at heating and cooling rates of 0.7 °C/s. The measurement accuracy was better than ±7 °C.

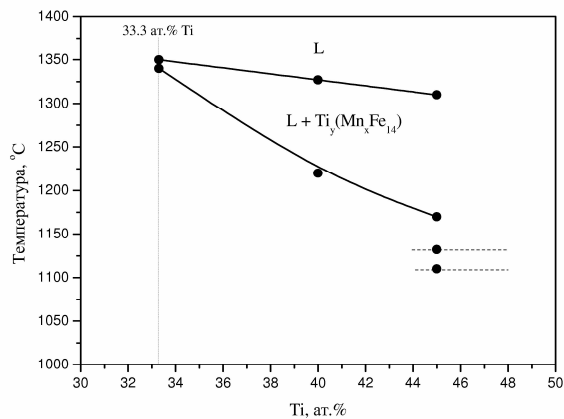
Experimental results and discussion

The DTA results are presented in **Table 1**. The corresponding fusion diagram is given in **Fig. 1**. It is of cigar type and resembles the diagram that can be constructed using the data of [1], but the liquidus temperatures are substantially lower than the values reported by [1]. The melting temperature of TiFe₂ was measured as 1440±7 °C. This value is 13 °C higher than the one accepted by [3,4], but 40 °C lower than the one reported by [6]. The melting temperature of TiMn₂ was measured as 1325±7 °C and coincides with the value accepted by [7].

**Fig. 1** Fusion diagram of the TiFe₂-TiMn₂ system.

The crystallographic cell parameters of TiMn₂ and TiFe₂ (MgZn₂ structure type) were measured as $a = 483.0$ pm, $c = 792.3$ pm, and $a = 479.2$ pm, $c = 781.4$ pm, respectively. The cell parameters as a function of the Fe content for the TiMn₂-TiFe₂ section obey the Vegard rule and may be presented as $a = 483.0 - 0.057X_{\text{Fe}} \pm 0.3$ pm, $c = 792.3 - 0.163X_{\text{Fe}} \pm 0.6$ pm, where X_{Fe} is at.% Fe in the alloy.

The substitution of Mn for Ti in the Laves phase with 14 at.% Fe (alloys #3, #8, #9) leads to a decrease of the solidus and liquidus temperatures, as shown in **Fig. 2**. But the heating curve of Ti₄₅Mn₄₁Fe₁₄ (**Fig. 3b**) differs from the one of Ti_{33.3}Mn_{52.7}Fe₁₄ (**Fig. 3a**) by the presence of two additional calorific effects. The comparison of the heating and cooling curves presented in **Fig. 3b** with that of the binary alloy Ti₅₀Mn₅₀ (**Fig. 4**) [8], shows that they are very similar and differ only by the temperatures of the phase transformations. Hence, the additional calorific effects on the heating curve in **Fig. 3b** may be explained by the reactions $(\beta\text{-Ti}) + \text{Ti}_v\text{Fe}_{14}\text{Mn}_z + \rho\text{-TiMn} \leftrightarrow \text{L} + \rho\text{-TiMn}$ (1110-1132 °C) and $\rho\text{-TiMn} \leftrightarrow \text{L} + \text{Ti}_v\text{Fe}_{14}\text{Mn}_x$ (1132-1170 °C). Melting of the Laves phase is observed in the temperature interval 1170–1310 °C.

**Fig. 2** The part of isopleth with 14 at.% Fe between 33.3 and 45 at.% Ti.

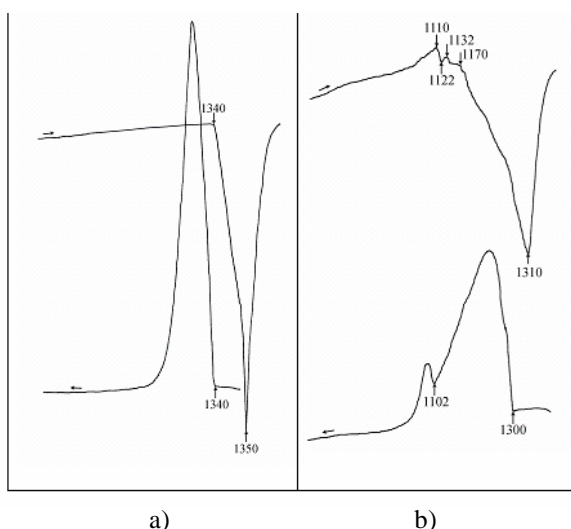


Fig. 3 Heating and cooling curves of the Ti_{33.3}Mn_{52.7}Fe₁₄ (a), and Ti₄₅Mn₄₁Fe₁₄ (b) alloys.

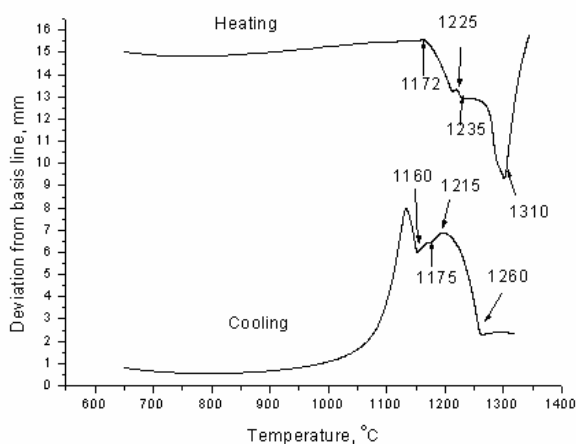


Fig. 4 Heating and cooling curves of the binary Ti_{0.5}Mn_{0.5} alloy.

The existence of a ρ -phase in the Ti₄₅Mn₄₁Fe₁₄ alloy does not confirm the statement in [4] that the Ti-rich boundary of the Ti(Mn,Fe)₂ phase field bends sharply towards the Ti-corner when the Fe content falls to approximately 14 at.%.

For the part of the isopleth with 14 at.% Fe that is located between 33.3 and 45 at.% Ti the crystallographic cell parameters as a function of the Ti content may be presented as $a = 457.7 + 0.731 X_{\text{Ti}} \pm \pm 0.5$ pm, $c = 763.06 + 0.791 X_{\text{Ti}} \pm 0.7$ pm, where X_{Ti} is at.% Ti in the alloy.

Conclusions

The polythermal section TiMn₂-TiFe₂ is quasibinary with a cigar-type fusion diagram. The melting points of TiMn₂ and TiFe₂ are 1325 ± 7 and 1440 ± 7 °C, respectively. The composition dependence of the cell parameters obeys the Vegard rule.

Since calorific effects caused by phase transformations with participation of a ρ -phase were observed in the Ti₄₅Mn₄₁Fe₁₄ alloy, the sharp bend of the Ti-rich boundary of the Ti(Mn,Fe)₂ field at 14 at.% Ti, reported earlier, is absent.

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