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- Summary and key words
- Introduction
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- Conclusions
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Main investigation results are revealed in Summary in brief (no more than 250 words).

Summary should include the following basic sections: topicality, statement of the problem, methods of settling the task, results, conclusions.

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All the designations and denominations of physical and chemical units of measurement should be given in SI system. Amino acids are designated by three letters.

The enzymes being used in the work, their recommended or systematic name and cipher should be given, following recommendations of International Biochemical Society (Enzyme Nomenclature – Acad. Press. San Diego. California and Supplement (1-6) in Eur. J. Biochem. (1993-1997, 1999) or electron version: http://www.chem.qmul.ac.uk/iubmb/enzyme.

The enzyme activity should be expressed through the rate of the catalyzed reaction in  $\mu M$  of the transformed substrate for 1 min per 1 mg of protein. Sometimes they use two more units of enzymatic activity: standard unit of activity U (IU) and catal (cat, in brief) simplifying the transition to SI system. Specific enzyme activity is usually expressed in  $\mu M/min$  per 1 mg of protein or in un.act./mg, cat/kg (R. M. C. Dawson, D. C. Elliott, W. H. Elliott, K. M. Jones. Data for Biochemical

Research. – 3th edition. Clarendon Press, Oxford, 1986). In all cases the reaction conditions are specified, i.e. temperature, pH, substrate concentration.

Solutions should be concentrated in M, mM,  $\mu$ M, etc., but not in normal concentration (n.). If concentration is expressed in percent, it should be specified, i.e., indices mass/mass, mass/volume, volume/volume should be mentioned. Salts used for making solutions — crystalline hydrates or waterless — should be also mentioned.

A term **relative molecular mass M**<sub>r</sub> (ratio between substance molecule mass to 1/12 of carbon atom  $C_{12}$ ), having no size, or term **molecular weight**  $(M_w)$  expressed in Da (Dalton) or in kDa are used for substances characteristics.

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#### Results and Discussion

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- Conclusions (without title)
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# **Examples of References**

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# Abbreviations for half-systematic and trivial names of chemical compounds

Nucleotides	
Adenosine-5'-mono-, di- and triphosphates	AMP, ADP, ATP
Guanisine-5'-mono-, di- and triphosphates	GMP, GDP, GTP
Ribothymisine-5'-mono-, di- and triphosphates	rTMP, rTDP, rTTP
Uridine-5'-mono-, di- and triphosphates	UMP, UDP, UTP
Cytidine-5'-mono-, di- and triphosphates	CMP, CDP, CTP
Designation of AMP isomers: 2'-AMP, 3'-AMP, 5'-AMP, 3':5'-AMP (ad	denosine-3':5'-monophosphate, cAMP.
Nucleic acids	
Desoxyribonucleic acid	DNA
Complementary DNA	cDNA
Mitochondrial DNA	mtDNA
Ribonucleic acid	RNA
Mitochondrial RNA	mtRNA
Matrix (information) RNA	mRNA
Ribosomal RNA	rRNA
Transport RNA	tRNA
tRNA with designation of acceptor specificity	tRNA <sup>Ala</sup> , tRNA <sub>2</sub> <sup>Ala</sup> , etc.
Aminoacylderivatives of tRNA	Ala-tRNA <sup>Ala</sup> , Glu-tRNA <sup>Glu</sup> , etc.
Others	
Diethylaminoethylenecellulose	DEAE-cellulose
Carboxymethylcellulose	CM-cellulose, CMC
Tris(hydroxymethyl) aminomethane	tris
Ethylendiaminetetraacetic acid (acetate)	EDTA
Ethylenglycoltetraacetic acid (acetate)	EGTA
Trichloroacetic acid	TCA
Polyacrylamide gel	PAAG
Coenzyme A	CoA
Sodium dodecylsulphate	SDS
Flavinadeninedinucleotide and its renewed form	FAD, FADH <sub>2</sub>
Nicotinamide adenine nucleotide, its oxidized and renewed forms	NAD, NAD <sup>+</sup> , NADH
Nicotinamide adenine nucleotidephosphate,	
its oxidized and renewed forms	NADP, NADP+, NADPH
Inorganic phosphate	P <sub>i</sub>
Inorganic pyrophosphate	PP <sub>i</sub>