A. V. Glushkov, P. A. Kondratenko, Yu. M. Lopatkin, V. V. Buyadzhi, A S. Kvasikova

<sup>1</sup>National Aviation University, 1, Komarov Ave, 03680 Kiev, Ukraine
 <sup>2</sup>Odessa National Polytechnical University, 1, Shevchenko av., Odessa, Ukraine
 <sup>3</sup>Odessa State Environmental University, 15, Lvovskaya str., Odessa, Ukraine
 e-mail: dirac13@mail.ru

# SPECTROSCOPY OF COOPERATIVE LASER ELECTRON- $\gamma$ NUCLEAR PROCESSES IN MULTIATOMIC MOLECULES: OsO<sub>4</sub>

The consistent quantum approach to calculating the electron-nuclear  $\gamma$  transition spectra (a set of the vibration-rotational satellites in a spectrum of molecule) of a nucleus in the multiatomic molecules is used to obtain the accurate data on the vibration-nuclear transition probabilities in a case of the emission and absorption spectrum of the nucleus <sup>188</sup>Os ( $E^{(0)}$ )  $_{\gamma}$  = 155 ke V) in the molecule of  $OsO_4$ . Analysis shows that more sophisticated theoretical approach gives the higher values for the cited probabilities.

### 1. Introduction

Any alteration of the molecular state must be manifested in the quantum transitions, for example, in a spectrum of the  $\gamma$ -radiation of a nucleus. It is well known that it is possible the transfer of part of a nuclear energy to atom or molecule under radiating (absorption) the  $\gamma$  quanta by a nucleus (c.f.[1-12]). A spectrum contains a set of the electron-vibration-rotation satellites, which are due to an alteration of the state of system interacting with photon. A mechanism of forming satellites in the molecule is connected with a shaking of the electron shell resulting from the interaction between a nucleus and  $\gamma$  quantum. [1,3,13-20]. An appearance of the molecular nuclear lines represents a great interest as it opens a possibility of the changing  $\gamma$ -radiation spectrum by means of changing the vibration-rotational state of a molecule by a laser light [1,12,16]. The well known example is the Szilard-Chalmers effect (molecular dissociation because of the recoil during radiating gamma quantum with large energy; c.f. [1, 2]). This paper is going on our studying the co-operative dynamical phenomena (c.f.[14-19]) due the interaction between atoms, ions, molecule

advanced approach to description of a new class of dynamical laser-electron-nuclear effects in molecular spectroscopy, in particular, we study the nuclear gamma-emission or absorption spectrum of a molecule. A consistent quantum- mechanical approach to calculation of the electron-nuclear  $\gamma$ transition spectra (set of vibration-rotational satellites in molecule) of a nucleus in the multiatomic molecules has been earlier proposed [14-16]. It generalizes the well known Letokhov-Minogin model [3]. Estimates of the vibration-nuclear transition probabilities in a case of the emission and absorption spectrum of nucleus <sup>191</sup>Ir ( $E^{(0)} = 82$ keV) in the molecule of  $IrO_4$  and other molecules were listed. Here we present the accurate data on the vibration-nuclear transition probabilities in a case of the emission and absorption spectrum of nucleus nucleus  $^{188}Os~(E^{(0)}_{\gamma}=$  155 keV) in the molecule of  $OsO_4$ .

electron shells and nuclei nucleons. We present an

# 2. The electron-nuclear $\gamma$ transition spectra of nucleus in multi-atomic molecules

As the method of computing is earlier presented in details [14-16], here we consider only by the key topics The main purpose is calculating a structure of the gamma transitions (a probability of transition) or spectrum of the gamma satellites because of changing the electron-vibration-rotational states of the multi-atomic molecules under gamma quantum radiation (absorption). Here we are limited by a case of the five-atomic molecules (of XY<sub>4</sub> type; T<sub>d</sub>). Hamiltonian of interaction of the gamma radiation with a system of nucleons for the first nucleus can be expressed through the co-ordinates of nucleons  $r_n$  in a system of the mass centre of the one nucleus [3,15]:

$$H(r_n) = H(r_n) \exp(-k_{\gamma} u) \tag{1}$$

where  $k_{\gamma}$  is a wave vector of the gamma quantum; u is the shift vector from equality state (coinciding with molecule mass centre) in system of co-ordinates in the space. The matrix element for transition from the initial state "a" to the final state "b" is presented as usually:

$$<\Psi_{b}^{*} \mid H \mid \Psi_{a} > \bullet < \Psi_{b}^{*} \mid e^{-ik_{y}u} \mid \Psi_{a} > \qquad (2)$$

where a and b is a set of quantum numbers, which define the vibrational and rotational states before and after interaction (with gamma- quantum). The first multiplier in eq. (2) is defined by the gamma transition of nucleus and is not dependent upon the internal structure of molecule in a good approximation. The second multiplier is the matrix element of transition of the molecule from the initial state "a" to the final state "b":

$$M_{ba} = \langle \Psi_{b}^{*}(r_{e}) | \Psi_{a}(r_{e}) \rangle \bullet$$
$$\bullet \langle \Psi_{b}^{*}(R_{1}, R_{2}) | e^{-ik_{\gamma}R_{1}} | \Psi_{a}(R_{1}, R_{2}) \rangle \qquad (3)$$

The expression (3) gives a general formula for calculating the probability of changing the internal state of molecule during absorption or emitting  $\gamma$  quantum by a nucleus. It determines an intensity of the corresponding  $\gamma$ -satellites. Their positions are fully determined as follows:

$$E_{\gamma} = E_{\gamma}^{0} \pm R + \hbar k_{\gamma} \nu \pm (E_{b} - E_{a}) \qquad (4)$$

Here M is the molecule mass, v is a velocity

of molecule before interaction of nucleus with  $\gamma$  quantum;  $E_a$  and  $E_b$  are the energies of the molecule before and after interaction;  $E_{\gamma}$  is an energy of nuclear transition;  $R_{om}$  is an energy of recoil:  $R_{om} = [(E_{\gamma}^{(o)}]^{2}/2Mc^{2}]$ . One can suppose that only single non-generated normal vibration (vibration quantum  $\hbar\omega$ ) is excited and initially a molecule is on the vibrational level  $v_a = 0$ . If we denote a probability of the corresponding excitation as  $P(v_b, v_a)$  and use expression for shift u of the  $\gamma$ -active nucleus through the normal co-ordinates, then an averaged energy for excitation of the single normal vibration is as follows:

$$\overline{E}_{vib} = \sum_{\nu=0}^{\infty} \hbar \omega \left( \nu + \frac{1}{2} \right) \overline{P}(\nu, 0) - \hbar \infty / 2 =$$

$$= \sum_{\nu=0}^{\infty} \hbar \omega \left( \nu + \frac{1}{2} \right) P(\nu, 0) - \hbar \omega / 2 =$$

$$\sum_{\nu=0} \hbar \omega \left( \nu + \frac{1}{2} \right) \frac{z^{\nu}}{\nu!} e^{-z} - \frac{\hbar \omega}{2} = \frac{1}{2} R \left( \frac{M - m}{m} \right), \quad (5)$$

where  $z = (R/\hbar\omega)[M-m/m]\cos^2 \vartheta$ , and m is the mass of  $\gamma$ -active nucleus,  $\vartheta$  is an angle between nucleus shift vector and wave vector of  $\gamma$ -quantum and line in  $\overline{E}_{vib}$  means averaging on orientations of molecule (or on angles  $\vartheta$ ).

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To estimate an averaged energy for excitation of the molecule rotation, one must not miss the molecule vibrations as they provide non-zeroth momentum  $L=k_v usin \vartheta$ , which is transferred to a molecule by  $\gamma$ -quantum. In supposing that a nucleus is only in the single non-generated normal vibration and vibrational state of a molecule is not changed  $v_a = v_b = 0$ , one could evaluate an averaged energy for excitation of the molecule rotations as follows:

$$\overline{E}_{\text{rot}} = \left\langle \overline{BL^2} \right\rangle = Bk_{\gamma}^2 \left\langle u^2 \right\rangle \overline{\sin^2 \vartheta} =$$
  
=  $\frac{1}{2} R(B/\hbar\omega) [(M-m)/m]$  (6)

As for multi-atomic molecules it is typical  $B/\hbar\omega \sim 10^{-4} \cdot 10^{-2}$ , so one could miss the molecule rotations and consider  $\gamma$ -spectrum of a nucleus in the molecule mass centre as a spectrum of the vibration-nuclear transitions. A shift u of the  $\gamma$ -active nucleus can be expressed through the normal co-ordinates  $Q_{sp}$  of a molecule:

$$u = \frac{1}{\sqrt{m}} \sum_{s\sigma} b_{s\sigma} Q_{s\sigma} \tag{7}$$

where *m* is a mass of the  $\gamma$ - active nucleus; components of the vector  $b_{s\sigma}$  of nucleus shift due to the  $\sigma$ -component of "s" normal vibration of a molecule are the elements of matrix *b* [2]; it realizes the orthogonal transformation of the normal co-ordinates matrix *Q* to matrix of masses of the weighted Cartesian components of the molecule nuclei shifts *q*. According to eq.(1), the matrix element can be written as multiplying the matrix elements on molecule normal vibration, which takes contribution to a shift of the  $\gamma$ - active nucleus:

$$M(b,a) = = \prod_{s} \left\langle v_{s}^{b} \mid \prod_{\sigma} \exp(-k_{\gamma} b_{s\sigma} Q_{s\sigma} / \sqrt{m}) v_{s}^{a} \right| \right\rangle. (8)$$

It is obvious that missing molecular rotations means missing the rotations which are connected with the degenerated vibrations. Usually wave functions of a molecule can be written for nondegenerated vibration as:

$$|v_{s}\rangle = \Phi_{s}(Q_{s}), \qquad (9)$$

for double degenerated vibration in the form:

$$|v_{s}\rangle = (v_{s}+1)^{-\frac{1}{2}} \sum_{s \sigma_{1}, s \sigma_{2}, s \sigma_{3}} \Phi_{v_{s\sigma_{1}}}(Q_{s\sigma_{1}}) \Phi_{v_{s\sigma_{2}}}(Q_{s\sigma_{2}}) (10)$$

(where  $v_{s\sigma_1} + v_{s\sigma_2} = v_s$ ) and for triple degenerated vibration as follows:

$$|v_{s}\rangle = \left(\frac{2}{(v_{s}+1)(v_{s}+2)}\right)^{1/2} \times \\ \times \sum_{s \sigma_{1}, s \sigma_{2}, s \sigma_{3}} \Phi_{v_{s\sigma_{1}}}(Q_{s\sigma_{1}}) \Phi_{v_{s\sigma_{2}}}(Q_{s\sigma_{2}}) \Phi_{v_{s\sigma_{3}}}(Q_{s\sigma_{3}})$$
(11)

where  $v_{s\sigma_1} + v_{s\sigma_2} + v_{s\sigma_3} = v_s$ .

In the simple approximation function  $\Phi_{v_{s\sigma}}(Q_{s\sigma})$  can be chosen in a form of the linear harmonic oscillator one. More exact calculating requires a numerical determination of these func-

tions. Taking directly the wave functions  $|v_s^a\rangle$  and  $|v_s^b\rangle$ , calculating the matrix element (8) is

reduced to a definition of the matrix elements on each component  $\gamma$  of the normal vibration.

#### 3. Results and conclusions

Below we present the accurate data on the vibration-nuclear transition probabilities in a case of the emission and absorption spectrum of nucleus <sup>188</sup>Os ( $E^{(0)} = 155$  keV in the molecule OsO<sub>4</sub>. Note that the main difficulty during calculating (8) is connected with definition of the values  $b_{sa}$ of the normalized shifts of  $\gamma$ -active decay. It is known that if a molecule has the only normal vibration of the given symmetry type, then the corresponding values of  $b_{s\sigma}$  can be found from the well known Eccart conditions, normalization one and data about the molecule symmetry. For several normal vibrations of the one symmetry type, a definition of  $b_{sy}$  requires solving the secular equation for molecule  $|GF - \lambda E| = 0$  [21-24]. We have used the results of advanced theoretical calculating electron structure of the studied system within an advanced relativistic scheme of the X<sub>-</sub> scattered waves method (see details in Refs. [21,24]). In table 1 we present the results of calculating probabilities of the first several vibration-nuclear transitions in a case of the emission and absorption spectrum of nucleus <sup>188</sup>Os ( $E_{\gamma}^{(0)} = 155$  keV in  $OsO_4$  and compare with available data Ref.[2], where more simple approximation has been used. Analysis shows that more sophisticated calculation gives the higher values for probabilities.

Table 1

Probabilitites of the vibrational-nuclear transitions in spectrum of OsO<sub>4</sub>.

Vibration transition $v_3^{\ a}, v_4^{\ a} - v_3^{\ b}, v_4^{\ b}$	$ \frac{\overline{P}(v_{3}^{a}, v_{4}^{a} - v_{3}^{b}, v_{4}^{b})}{[15]} $	$\overline{P}\left(v_{3}^{a}, v_{4}^{a} - v_{3}^{b}, v_{4}^{b}\right)$ This work
0,0 - 0,0	0.795	0.803
1,0 - 0,0	0.018	0.023
0,1 - 0,0	0.074	0.081
1,0 - 1,0	0.750	0.757
0,1 - 0,1	0.673	0.678

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# SPECTROSCOPY OF COOPERATIVE LASER ELECTRON-G-NUCLEAR PROCESSES IN MULTIATOMIC MOLECULES: $OsO_4$

### Abstract

The consistent quantum approach to calculating the electron-nuclear g transition spectra (a set of the vibration-rotational satellites in a spectrum of molecule) of a nucleus in the multiatomic molecules is used to obtain the accurate data on the vibration-nuclear transition probabilities in a case of the emission and absorption spectrum of the nucleus <sup>188</sup>Os ( $E^{(0)} = 155$  keV) in the molecule of  $OsO_4$ . Analysis shows that more sophisticated theoretical approach gives the higher values for the cited probabilities.

Key words: electron-g-nuclear transition spectrum, multiatomic molecules

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А. В. Глушков, П. А. Кондратенко, Ю. М. Лопаткин, В. В. Буяджи, А. С. Квасикова

# СПЕКТРОСКОПИЯ КООПЕРАТИВНЫХ ЛАЗЕРНЫХ ЭЛЕКТРОННО-ГАММА-ЯДЕРНЫХ -ПРОЦЕССОВ В МНОГОАТОМНЫХ МОЛЕКУЛАХ: OsO<sub>4</sub>

#### Резюме

Последовательный квантовый подход к расчету электронно-гамма-ядерного спектра (система колебательно-вращательных спутников в спектре молекуле) в многоатомных молекулах используется, чтобы получить уточненные данные о вероятностях колебательно-ядерных переходов в молекуле  $OsO_4$  в случае испускания и поглощения гамма-кванта ядром <sup>188</sup>Os ( $E^{(0)}_{g}$  = 155 kэB). Анализ показывает, что более последовательный теоретический подход дает более высокие значения искомых вероятностей.

Ключевые слова: спектр электрон- g -ядерных переходов, многоатомные молекулы

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О. В. Глушков, П. О. Кондратенко, Ю. М. Лопаткін, В. В. Буяджи, Г. С. Квасикова

# СПЕКТРОСКОПІЯ КООПЕРАТИВНИХ ЛАЗЕРНИХ ЕЛЕКТРОННО-ГАММА-ЯДЕРНИХ ПРОЦЕСІВ У БАГАТОАТОМНИХ МОЛЕКУЛАХ: OsO<sub>4</sub>

## Резюме

Послідовний квантовий підхід до розрахунку електронно-гамма-ядерного спектру (система коливально-обертальних супутників у спектрі молекули) в багатоатомних молекулах використовується, щоб отримати уточнені дані по ймовірностям коливально-ядерних переходів в молекулі  $OsO_4$  у випадку випромінювання та поглинання гамма-кванта ядром <sup>188</sup>Os ( $E^{(0)}_{g}$  = 155 кеВ). Аналіз показує, що більш послідовний теоретичний підхід дає більш високі значення шуканих ймовірностей.

Ключові слова: спектр електрон- д -ядерних переходів, багатоатомні молекули